

configuration, as the other possibility, H-6 and H-8 α -orientated, did not explain the observed shifts. Consequently, the new diterpene was 7-oxo-6 α -hydroxy-hardwickiic acid lactone (**1**). Obviously, further species of *Pulicaria* will have to be investigated to clarify the chemotaxonomy of this genus. So far relationships to *Inula* can be stated though this genus does not appear very uniform with regard to its chemistry.

EXPERIMENTAL

The air-dried plant material (250 g), collected near Teheran (voucher A. R. 115) was extracted with Et₂O-petrol (1:2). CC (Si gel) and TLC (Si gel) of the polar fractions (Et₂O-CH₂Cl₂, 2:1) gave 30 mg colourless crystals (**1**), mp 161°. IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1785, 1750, 1690, 1605, 1515, 885; MS m/z (rel. int.): 328.167 [M]⁺ (42), 313 [M - Me]⁺ (8), 233 [M - CH₂CH₂ furane]⁺ (19), 205

[233 - CO]⁺ (20), 95 [⁺CH₂CH₂ furane]⁺ (91), 81 [pyrilium]⁺ (100);

$$[\alpha]_{24}^D = \frac{589}{-62.3} \quad \frac{578}{-65.0} \quad \frac{546 \text{ nm}}{-74.9} \quad (c = 2.39, \text{CHCl}_3).$$

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REFERENCES

1. Bohlmann, F., Burkhardt, T. and Zdero, C. (1973) *Naturally Occurring Acetylenes*, p. 355. Academic Press, New York.
2. Bohlmann, F., Knoll, K.-H. and El-Emary, N. A. (1979) *Phytochemistry* **18**, 1231.
3. Schulte, K. E., Reisch, J. and Hopman, J. (1963) *Arch. Pharm.* **296**, 353.
4. Schulte, K. E., Reisch, J. and Müller, F. (1968) *Arch. Pharm.* **301**, 115.

GUAIANOLIDES FROM *CENTAUREA CANARIENSIS**

FERDINAND BOHLMANN and RAJINDER K. GUPTA

Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, West Germany

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Key Word Index—*Centaurea canariensis*; Compositae; sesquiterpene lactones; guaianolides; costic acid derivative.

Abstract—The aerial parts of *Centaurea canariensis* afforded four new guaianolides, all closely related to dehydrocostus lactone, and a derivative of costic acid. The structures were elucidated by spectroscopic methods.

INTRODUCTION

Many species from the large genus *Centaurea* (tribe Cynareae) have been investigated chemically. In addition to polyacetylenes [1] several sesquiterpene lactones [2] have been isolated. *C. canariensis* Brouss. var. *subexpinnata* Burch. also contains acetylenes and several sesquiterpene lactones, all closely related to dehydrocostus lactone.

RESULTS AND DISCUSSION

The roots of *C. canariensis* afforded the tetrayne **1** and aptotaxene (**2**), while the aerial parts gave capillol acetate

(**3**) [1], germacrene D, bicyclogermacrene, γ - and δ -cadinene and five new compounds, the costic acid derivative **4** and four guaianolides (**7–10**). **4** after addition of diazomethane afforded on oxidation the ketone **6**. The ¹H NMR data of **4** and **6** (Table 1) clearly showed that **4** was a derivative of costic acid. Spin decoupling allowed the assignment of all signals. The position of the hydroxy group followed from the ¹H NMR data of the corresponding ketone **6**. The downfield shift of the broadened singlets of H-15 agreed only with a keto group at C-3. The small coupling $J_{2,3}$ required a 3 β -hydroxy group. The stereochemistry at C-5 and C-7 followed from the couplings observed for H-5 and H-7. Consequently, the ¹H NMR data were in part very similar to those of costic acid. The ¹H NMR data of **7**, **9** and **10** were close to those of 8 α -seneciolyloxy dehydrocostus lactone [3]. (Table 1). Spin decoupling allowed the assignment of all

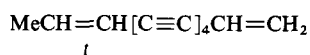
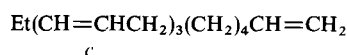
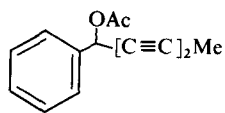
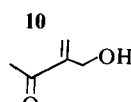
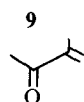
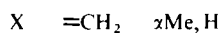
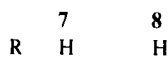
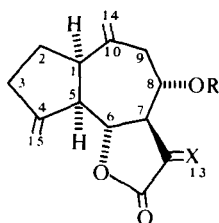
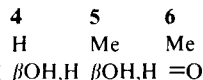
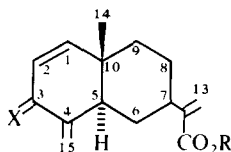
* Part 372 in the series 'Naturally Occurring Terpene Derivatives'. For Part 371, see Bohlmann, F. and Zdero, C. (1981) *Phytochemistry* **20** 2529.

Table 1. ^1H NMR spectral data of compounds **4** and **6–10** (400 MHz, CDCl_3 , TMS as internal standard)

	4 (C ₆ D ₆)	6	7	8*	9	10
H-1	5.40 <i>dd</i>	6.00 <i>d</i>	2.99 <i>ddd</i>	2.93 <i>ddd</i>	2.99 <i>ddd</i>	3.00 <i>ddd</i>
H-2 α	5.47 <i>dd</i>	6.85 <i>d</i>	1.83 <i>m</i>	1.91 <i>m</i>	1.84 <i>m</i>	1.83 <i>m</i>
H-2 β				1.85 <i>m</i>		
H-3 α	5.32 <i>s</i> (<i>br</i>)	—	2.44 <i>m</i>	2.48 <i>m</i>	2.45 <i>m</i>	2.45 <i>dddd</i>
H-3 β			2.51 <i>m</i>	2.54 <i>m</i>	2.54 <i>m</i>	2.53 <i>m</i>
H-5	2.01 <i>d</i> (<i>br</i>)	1.88 <i>d</i> (<i>br</i>)	2.81 <i>dd</i> (<i>br</i>)	2.81 <i>dd</i> (<i>br</i>)	2.82 <i>dd</i> (<i>br</i>)	2.83 <i>dd</i> (<i>br</i>)
H-6 α	1.69 <i>d</i> (<i>br</i>)	1.77 <i>d</i> (<i>br</i>)	—	—	—	—
H-6 β	1.41 <i>dddd</i>	1.62 <i>m</i>	3.96 <i>dd</i>	3.89 <i>dd</i>	4.05 <i>dd</i>	4.04 <i>dd</i>
H-7	2.56 <i>dddd</i> (<i>br</i>)	2.64 <i>dddd</i> (<i>br</i>)	2.83 <i>dddd</i>	2.48 <i>m</i>	3.21 <i>dddd</i>	3.23 <i>dddd</i>
H-8 α	1.59 <i>dd</i> (<i>br</i>)	1.75 <i>m</i> 1.6 <i>m</i> 1.5 <i>m</i>	—	—	—	—
H-8 β	1.34 <i>m</i>		3.92 <i>ddd</i>	3.71 <i>m</i>	5.07 <i>ddd</i>	5.01 <i>ddd</i>
H-9 α	1.4 <i>m</i>		2.69 <i>dd</i>	2.70 <i>dd</i>	2.70 <i>dd</i>	2.72 <i>dd</i>
H-9 β			2.26 <i>dd</i>	2.14 <i>dd</i>	2.34 <i>dd</i>	2.35 <i>dd</i>
H-13	6.34 <i>s</i> (<i>br</i>)	6.22 <i>s</i> (<i>br</i>)	6.27 <i>d</i>	1.40 <i>d</i>	6.23 <i>d</i>	6.22 <i>d</i>
H-13'	5.36 <i>s</i> (<i>br</i>)	5.62 <i>s</i> (<i>br</i>)	6.15 <i>d</i>		5.62 <i>d</i>	5.62 <i>d</i>
H-14	0.79 <i>s</i>	0.96 <i>s</i>	5.05 <i>s</i> (<i>br</i>)	4.94 <i>s</i> (<i>br</i>)	5.04 <i>s</i> (<i>br</i>)	5.05 <i>s</i> (<i>br</i>)
H-14'			4.93 <i>s</i> (<i>br</i>)	4.91 <i>s</i> (<i>br</i>)	4.92 <i>s</i> (<i>br</i>)	4.91 <i>s</i> (<i>br</i>)
H-15	5.32 <i>s</i> (<i>br</i>)	6.08 <i>s</i> (<i>br</i>)	5.28 <i>s</i> (<i>br</i>)	5.22 <i>s</i> (<i>br</i>)	5.29 <i>s</i> (<i>br</i>)	5.28 <i>s</i> (<i>br</i>)
H-15'	4.69 <i>s</i> (<i>br</i>)	5.17 <i>s</i> (<i>br</i>)	5.07 <i>s</i> (<i>br</i>)	5.04 <i>s</i> (<i>br</i>)	5.09 <i>s</i> (<i>br</i>)	5.04 <i>s</i> (<i>br</i>)
OCOR	—	—	—	—	6.18 <i>s</i> (<i>br</i>)	6.33 <i>s</i> (<i>br</i>)
					5.67 <i>s</i> (<i>br</i>)	5.96 <i>s</i> (<i>br</i>)
					1.95 <i>s</i> (<i>br</i>)	4.38 <i>s</i> (<i>br</i>)
OMe	—	3.77 <i>s</i>	—	—	—	—

* H-11, *dq* 2.54.

$J(\text{Hz})$: compound **4**: 1,2 = 10; 1,3 = 1.5; 2,3 = 2.5; 5,6 β = 6 α ,6 β = 6 β ,7 α = 12; 6 α ,7 α = 7 α ,8 α = 3; 7 α ,8 β = 12; 8 α ,8 β = 13; 8 α ,9 α = 4; compound **6**: 1,2 = 10; 5,6 β = 6 α ,6 β = 6 β ,7 α = 12; 6 α ,7 α = 7 α ,8 α = 3; compounds **7–10**: 1,2 α = 1,2 β = 1,5 = 8.5; 5,6 = 10; 6,7 = 9; 7,13 = 3.5; 7,13' = 3.0; 7,8 = 10; 8,9 α = 5; 8,9 β = 4.5 (compound **8**: 7,8 β = 8; 7,11 = 10.5; 11,13 = 7; compound **10**: 3 α ,3 β = 17; 2 α ,3 α = 2 β ,3 α = 8.5; 3 α ,15 = 2).

**1****2****3**

signals, especially those of the main constituent **10**, where only a few signals were overlapped multiplets. The stereochemistry followed from the couplings observed for H-1, H-5, H-7 and H-8, while the nature of the oxygen function at C-8 was easily deduced from the typical ^1H NMR signals. In the ^1H NMR spectrum of **8** (Table 1) only a few signals were slightly different from those of **7**. However, the missing signals of methylene protons (H-13) were replaced by a methyl doublet at δ 1.40 and a doublet quartet at δ 2.54, indicating the presence of a dihydro derivative of **7**. The stereochemistry at C-11 followed from the large coupling $J_{7,11}$. **10** is a 3-desoxy derivative of cynaropicrin, which has been reported several times from the genus *Centaurea* [4–8]. Highly oxygenated germacranolides, however, are much more widespread [2].

EXPERIMENTAL

The air-dried plant material (collected from Tenerife, voucher 80/1480, deposited in the Herbarium of the Institute of Plant systematic, Göttingen) was extracted with Et_2O –petrol (1:2) and the resulting extracts were sepd by CC (Si gel) and repeated TLC (Si gel). Known compounds were identified by comparison of their IR and ^1H NMR spectra with those of authentic material. The roots (10 g) afforded 2 mg **1** and 10 mg **2**, while the aerial parts (200 g) gave 50 mg germacrene-D, 50 mg bicyclogermacrene, 5 mg γ - and 20 mg δ -cadinene, 5 mg **3**, 3 mg **4** (Et_2O –petrol, 3:1), 2 mg **7** (Et_2O –petrol, 1:1), 2 mg **8** (Et_2O –petrol, 1:1), 2 mg **9** (Et_2O –petrol, 1:1) and 20 mg **10** (Et_2O –petrol, 3:1).

3-Oxo-1,2-dehydrocostic acid (4). Colourless gum, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} : 3580 (OH), 3500–2700, 1690, 1620 ($\text{C}=\text{CCO}_2\text{H}$); MS m/z (rel. int.): 248.141 $[\text{M}]^+$ (81) ($\text{C}_{15}\text{H}_{20}\text{O}_3$), 230 $[\text{M} - \text{H}_2\text{O}]^+$ (24), 215 $[\text{230} - \text{Me}]^+$ (32), 202 $[\text{230} - \text{CO}]^+$ (30), 187 $[\text{202} - \text{Me}]^+$ (37), 91 $[\text{C}_7\text{H}_7]^+$ (100);

$$[\alpha]_{24}^{25} = \frac{589}{+18} + \frac{578}{+19} + \frac{546}{+22} + \frac{436}{+23} \text{ nm} (c = 0.15, \text{CHCl}_3).$$

To 3 mg **4** in 1 ml Et_2O was added excess CH_3N_2 . TLC (Et_2O –petrol, 1:1) afforded 3 mg **5**, which was stirred with 50 mg MnO_2 in 2 ml Et_2O for 2 hr. TLC (Et_2O –petrol, 1:3) afforded 2 mg **6**, colourless gum; MS m/z (rel. int.): 260.141 $[\text{M}]^+$ (53) ($\text{C}_{16}\text{H}_{20}\text{O}_3$), 245 $[\text{M} - \text{Me}]^+$ (37), 228 $[\text{M} - \text{MeOH}]^+$ (56), 213 $[\text{228} - \text{Me}]^+$ (23), 200 $[\text{228} - \text{CO}]^+$ (51), 185 $[\text{200} - \text{Me}]^+$ (47), 91 $[\text{C}_7\text{H}_7]^+$ (100).

8 α -Hydroxydehydrocostus lactone (7). Colourless crystals, mp 106° (Et_2O –petrol), IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3600 (OH), 1770 (γ -

lactone); MS m/z (rel. int.): 246.126 $[\text{M}]^+$ (5) ($\text{C}_{15}\text{H}_{18}\text{O}_3$), 228 $[\text{M} - \text{H}_2\text{O}]^+$ (38), 91 (78), 69 (100);

$$[\alpha]_{24}^{25} = \frac{589}{+76} + \frac{578}{+84} + \frac{546}{+100} + \frac{436}{+156} \text{ nm} (c = 0.1, \text{CHCl}_3).$$

8 α -Hydroxy-11 β ,13H-dehydrocostus lactone (8). Colourless gum, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} : 3620 (OH), 1790 (γ -lactone); MS m/z (rel. int.): 248.149 $[\text{M}]^+$ (18) ($\text{C}_{15}\text{H}_{20}\text{O}_3$), 230 $[\text{M} - \text{H}_2\text{O}]^+$ (21), 202 $[\text{230} - \text{CO}]^+$ (10), 158 $[\text{202} - \text{CO}_2]^+$ (63), 107 (100), 91 (79), 79 (85), 71 (85), 69 (63); $[\alpha]_{\text{D}} = +16^\circ (c = 0.1, \text{CHCl}_3)$.

8 α -Methacryloyloxy dehydrocostus lactone (9). Colourless gum, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} : 1790 (γ -lactone), 1730 ($\text{C}=\text{CCO}_2\text{R}$); MS m/z (rel. int.): 314.152 $[\text{M}]^+$ (0.3) ($\text{C}_{19}\text{H}_{22}\text{O}_4$), 228 $[\text{M} - \text{RCO}_2\text{H}]^+$ (28), 213 $[\text{228} - \text{Me}]^+$ (2), 200 $[\text{228} - \text{CO}]^+$ (9), 69 $[\text{C}_3\text{H}_5\text{CO}]^+$ (100);

$$[\alpha]_{24}^{25} = \frac{589}{+102} + \frac{578}{+110} + \frac{546}{+128} + \frac{436}{+165} \text{ nm} (c = 0.1, \text{CHCl}_3).$$

3-Desoxycynaropicrin (10). Colourless gum, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} : 3500 (OH), 1770 (γ -lactone), 1720, 1640 ($\text{C}=\text{CCO}_2\text{R}$); MS m/z (rel. int.): 330 $[\text{M}]^+$ (0.1), 312 $[\text{M} - \text{H}_2\text{O}]^+$ (0.2), 228.115 $[\text{M} - \text{RCO}_2\text{H}]^+$ (100) ($\text{C}_{15}\text{H}_{16}\text{O}_2$), 213 $[\text{228} - \text{Me}]^+$ (7), 200 $[\text{228} - \text{CO}]^+$ (7), 85 $[\text{HOC}_3\text{H}_4\text{CO}]^+$ (20);

$$[\alpha]_{24}^{25} = \frac{589}{+112} + \frac{578}{+117} + \frac{546}{+133} + \frac{436}{+183} \text{ nm} (c = 2.0, \text{CHCl}_3).$$

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REFERENCES

1. Bohlmann, F., Burkhardt, T. and Zdero, C. (1973) *Naturally Occurring Acetylenes*, p. 452. Academic Press, New York.
2. Fischer, N. H., Olivier, E. J. and Fischer, H. D. (1979) *Prog. Chem. Nat. Prod.* **38**, 48.
3. Bohlmann, F., Brindöpke, G. and Rastogi, R. C. (1978) *Phytochemistry* **17**, 475.
4. Ohno, N., Hirai, H., Yoshioka, H., Dominguez, X. A. and Mabry, T. J. (1973) *Phytochemistry* **12**, 221.
5. Gonzales, G. A., Bermejo, J. and Massanet, G. (1977) *Rev. Latinoam. Quim.* **8**, 176.
6. Gonzales, G. A., Bermejo, J., Cabrera, I., Massanet, G., Mansila, H. and Galindo, A. (1978) *Phytochemistry* **17**, 955.
7. Gonzales, A. G., Darias, V., Alonso, G. and Estevez, E. (1980) *Plant. Med.* **40**, 179.
8. Rustaiyan, A., Niknejad, A., Zdero, C. and Bohlmann, F. (1981) *Phytochemistry* **20** (in press).